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Key indicators

Single-crystal X-ray study T = 273 K Mean σ (C–C) = 0.004 Å R factor = 0.060 wR factor = 0.178 Data-to-parameter ratio = 19.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound, $C_{16}H_{17}O_5P$, was obtained by the reaction of benzoyl chloride and dimethyl (hydroxyphenylmethyl)phosphonate. The molecular packing is stabilized by van der Waals forces.

(Dimethoxyphosphoryl)(phenyl)methyl benzoate

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Comment

The title compound, (I), was synthesized for a study of its antimicrobial activity against *Bacillus subtilis*. The hydroxy-phosphonic acid derivative was found to have moderate antimicrobial activity (inhibition zone = 7.5 mm), The crystal packing is stabilized by van der Waals forces, and bond lengths and angles in (I) are in agreement with the values reported in the literature (Smaardijk *et al.*, 1985; Deron *et al.*, 2002).



Experimental

(Hydroxyphenylmethyl)phosphonic acid dimethyl ester, $C_6H_5CH_{(OH)P(O)(OCH_3)_2}$, was prepared according to the procedure of Fang *et al.* (2006). This ester (1.08 g, 5 mmol) was dissolved in dry tetrahydrofuran (20 ml) to which triethylamine (0.7 ml) was added, and the solution was added dropwise to benzoyl chloride (0.7 g, 5 mmol) in the same solvent (10 ml). After completion of the reaction, the precipitate was separated and the filtrate was extracted with ethyl acetate, dried over anhydrous MgSO₄ and concentrated under vacuum. The residual liquid was purified by column chromatography to give the title compound. Single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of a petroleum etherethyl acetate solution (10:1 v/v).

Crystal data $C_{16}H_{17}O_5P$ $M_r = 320.27$ Orthorhombic, *Pbca* a = 9.400 (2) Å b = 10.739 (2) Å c = 31.821 (7) Å V = 3212.2 (12) Å³

Z = 8 D_x = 1.324 Mg m⁻³ Mo K α radiation μ = 0.19 mm⁻¹ T = 273 (2) K Chunk, colorless 0.42 × 0.19 × 0.11 mm

Data collection

Bruker APEX area-detector diffractometer φ scans Absorption correction: multi-scan (*SADABS*; Bruker, 2001) $T_{\min} = 0.924, T_{\max} = 0.979$ 18166 measured reflections 3871 independent reflections 2947 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.070$ $\theta_{\text{max}} = 28.3^{\circ}$

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Refinement

Refinement on F^2	<i>w</i> =
$R[F^2 > 2\sigma(F^2)] = 0.060$	
$wR(F^2) = 0.178$	w
S = 1.04	$(\Delta/\sigma$
3871 reflections	$\Delta \rho_{\rm m}$
199 parameters	$\Delta \rho_{\rm m}$
H-atom parameters constrained	

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0984P)^{2} + 0.4253P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3 (\Delta/\sigma)_{max} = 0.001 \Delta\rho_{max} = 0.42 \text{ e} \text{ Å}^{-3} \Delta\rho_{min} = -0.41 \text{ e} \text{ Å}^{-3}$

All H atoms were placed in geometrically idealized positions and treated as riding on their parent atoms, with C–H = 0.93 (aromatic), 0.98 (CH) or 0.96 Å (CH₃), and U_{iso} (H) = $1.2U_{eq}$ (aromatic C, CH) or $1.5U_{eq}$ (methyl C).

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001; data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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References

- Bruker (2001). SAINT (Version 6.22), SMART (Version 5.625) and SADABS (Version 2.03). Bruker AXS Inc., Madison, Wisconsin, USA.
- Deron, A., Milewska, M., Barycki, J., Sawka-Dobrowolska, W. & Gancarz, R. (2002). Heteroat. Chem. 13, 157–164.
- Fang, H., Fang, M.-J., Zeng, Z.-P., Wei, Z.-B. & Zhao, Y.-F. (2006). Acta Cryst. E62, 01378–01379.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.



Figure 1

View of the molecular structure of (I) with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Smaardijk, A. A., Noorda, S., van Bolhuis, F. & Wynberg, H. (1985). Tetrahedron Lett. 26, 493–496.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.